S/048/59/023/011/009/012 B006/B056

AUTHOR:

Trofimov. A. K.

TITLE:

The Use of Luminescence for the Investigation of Reactions

in Solid Phases in the System CaO4-A12O3

PERIODICAL:

Izvestiya Akademii nauk SSSR. Seriya fizicheskaya, 1959,

Vol. 23, No. 11, pp. 1355-1359

TEXT: The author already gave a report on the crystallochemical investigations he carried out previously (Ref. 1); because of the great sensitivity of luminescence line spectra of rare earth- or chromophosphors to variations of the crystal structure, crystallochemical transformations may variations of the crystal structure, crystallochemical transformations may be well observed by means of the luminescence spectra. These investigations are mainly confined to determinations of the temperature limits of the existence of the individual solid phases. The spectra are, however, also existence of the individual solid phases in complicated compositions of suited for the observation of processes in complicated compositions of solid phases and the identification of the chemical composition of the individual phases. Arbitrary crystal-structural processes may be followed individual phases. Arbitrary crystal-structural processes may be followed which had already been investigated by other methods, and comparisons may which had already been investigated by other methods, and comparisons may be drawn. For this purpose, the author chose the system CaO-Al<sub>2</sub>O<sub>3</sub>, which Card 1/4

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The Use of Luminescence for the Investigation S/048/59/023/011/009/012 of Reactions in Solid Phases in the System B006/B056  ${\rm CaO_A-Al_2O_3}$ 

is of especial importance for cement production. By means of the luminescence line spectrum the author investigated the complex chemical reaction CaO and Al<sub>2</sub>O<sub>3</sub>, where Eu and Sm were used as activators. The investigations were carried out in the solid phase and with excitation by means of a GOI-phosphoroscope; recordings were made by means of a diffraction spectrograph (dispersion of the devices: 50 and 25 A/mm). A total of 36 original spectra are shown in four figures. They are discussed in all details in the following. Fig. 1 shows: CaO - Eu, Al<sub>2</sub>O<sub>3</sub> - Eu, CaO - Sm, and Al<sub>2</sub>O<sub>3</sub> - Sm; Fig. 2: 3 CaO·Al<sub>2</sub>O<sub>3</sub> - Eu, CaO·Al<sub>2</sub>O<sub>3</sub> - Eu, CaO·Al<sub>2</sub>O<sub>3</sub> - Eu, CaO·6Al<sub>2</sub>O<sub>3</sub> - Eu, Al<sub>2</sub>O<sub>3</sub> - Eu, and CaO - Eu, annealed at 1400° during one hour; Fig. 3: mixtures of various ratios of CaO + Al<sub>2</sub>O<sub>3</sub> (annealed at 1400°) - Table 2 shows the composition for 18 investigated mixtures. Fig. 4: CaO·6Al<sub>2</sub>O<sub>3</sub> annealed for 1 hour at 8 various temperatures between 700 and 1400°, and, for comparison, again the Al<sub>2</sub>O<sub>3</sub>-Eu and CaO-Eu spectra. Table 1 shows the exact spectral analysis of Fig. 2 - the wave lengths and the intensity ratios of the calcium aluminate lines investigated. From the lines found it is Card 2/4

The Use of Luminescence for the Investigation S/048/59/023/011/009/012 of Reactions in Solid Phases in the System B006/B056  $^{\text{CaO}}4^{-\text{Al}}2^{\text{O}}3$ 

possible to draw conclusions as to the manner in which the reactions develop. In detail, the following reactions are assumed: 1)  $3\text{CaO} \cdot \text{Al}_2\text{O}_3$  occurs at  $1200^\circ$  in a  $3\text{CaO} + \text{Al}_2\text{O}_3$  mixture, at lower temperatures  $\text{CaO} \cdot \text{Al}_2\text{O}_3$  is produced, and the following reaction develops:  $\text{CaO} \cdot \text{Al}_2\text{O}_3$  +  $2\text{CaO} \longrightarrow 3\text{CaO} \cdot \text{Al}_2\text{O}_3$ . 2) No  $5\text{CaO} \cdot 3\text{Al}_2\text{O}_3$  was found up to  $1400^\circ$ .

- 3) CaO-Al203 forms very rapidly within the range of 900 to 1000°.
- 4) Ca0.2A1,03 occurs at 1000°: Ca0.A1,03 + A1,03 -> Ca0.2A1,03.
- 5)  $\text{Ca0.6Al}_2\text{O}_3$  occurs at  $1100^\circ$ :  $\text{Ca0.2Al}_2\text{O}_3 + 4\text{Al}_2\text{O}_3 \rightarrow \text{Ca0.6Al}_2\text{O}_3$  or  $\text{Ca0.Al}_2\text{O}_3 + 5\text{Al}_2\text{O}_3 \rightarrow \text{Ca0.6Al}_2\text{O}_3$ . The author finally thanks P. P. Feofilov for his supervision and friendly assistance in carrying out

the investigation. There are 4 figures, 2 tables, and 7 references: 4 Soviet.

Card 3/4

The Use of Luminescence for the Investigation S/048/59/023/011/009/012 of Reactions in Solid Phases in the System B006/B056  ${\rm Ca0}_4{\rm -Al}_2{\rm O}_3$ 

ASSOCIATION:

Gos. opticheskiy institut im. S. I. Vavilova (State Optics Institute imeni S. I. Vavilov)

Card 4/4

TROFIMOV, A. K., Cand Chem Sci - (diss) "Solution of some chemical problems by means of observation of ruled spectra of the luminescence of crystal phosphors." /Leningrad/, 1960. 13 pp; (State Order of Lenin Optical Institute im S. I. Vavilova); 150 copies; price not given; (KL, 19-60, 130)

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ARAPOVA, E.Ya.; BARANOVA, Ye.G.; LEVSHIN, V.L.; TIMOFEYEVA, T.V.; TROFIMOV,
A.K.; FROFILOV, P.P.

Imminescent method of quantitative determination of gadolinium in
methallic beryllium. Trudy Kom. anal. khim. 12:344-354 '60.

(MIRA 13:8)

(Beryllium--Analysis)

(Gadolinium earths)

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S/048/61/025/004/007/048 B104/B201

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AUTHOR:

Trofimov, A. K.

TITLE:

Luminencence of lanthanides in thorium oxide

PERIODICAL:

Izvestiya Akademii nauk SSSR. Seriya Tizicheskaya, v. 25.

no. 4, 1961, 460-461

TEXT: The present paper has been read at the 9th Conference on Luminescence (Crystal Phosphors), Kiyev, June 20-25, 1960. The microchemical luminescence determination of rare earth elements consists in observing luminescence of crystal phosphors containing the said elements as the luminescence of crystal phosphors containing the said elements as activators. Thorium oxide is indicated as the most important compound suited for studies of this kind. Very little work has been, however, done in this direction, and the lack of data regarding the emission spectrum, in this direction, and the lack of data regarding the emission spectrum, the limiting concentration of the activator, etc., is felt. The author wanted to study the luminescence spectrum of trivalent lanthanides in wanted to study the luminescence spectrum of trivalent lanthanides in thorium oxide, and to clarify the problem, as to whether these elements might be usable in analytical chemistry. The luminescence of these phosphores is excited by shortwave UV; an arc phosphoroscope of the type FOM

Card 1/3

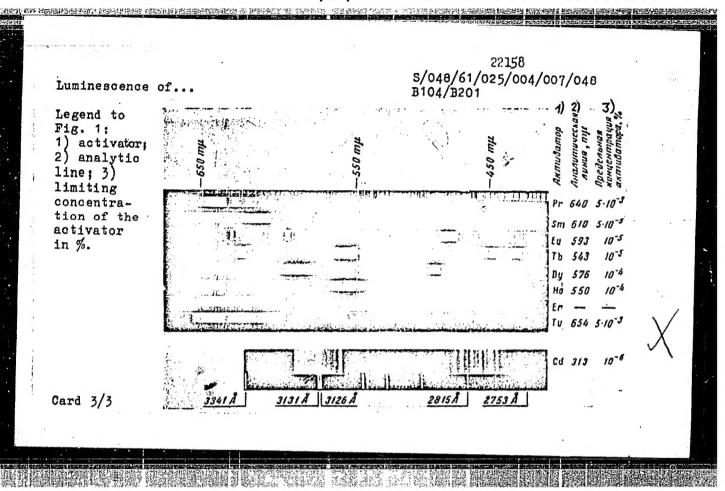
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S/048/61/025/004/007/048 B!04/B201

Luminescence of ...

(GOI) was used to observe the luminescence in any spectral region, without having to use light filters. The luminescence of Gd is in the ultraviolet, that of Pr, Sm, Eu, Tb, Dy, Ho, Er, and Th, and also of ThO2 is in the visible spectral region. Lines suited for the analysis were established in almost all of the elements. In Er, Ho, and Tb strong groups of lines are situated at nearly the same place in the spectrum. The lower limiting concentrations are presented in the figure, the upper is for all at 0.1 %. The emission of Nd and Yb is in the infrared region, and is therefore not suited for photographic methods. Crystalline thorium oxide is not only a good base for rare-earth phosphors, but also a good acceptor for the lanthanide ions. The effect of directional thermal diffusion has been worked out by the author for the quantitative determination of gadolinium in metallic beryllium (Ref. 7: Arapova, E. Ya. et al., Tr. Komissii po analit. khimii, 12. 344, (1960)). This method also permits the micro-chemical determination of lanthanides in minerals. P. P. Feofilov is thanked for his aid in the measurements. There are 1 figure and 7 references: 5 Soviet-bloc and 2 non-Soviet-bloc. The two references to English-language publications read as follows: Ref. 1: Nichols E. L., Wick F. G., J. Opt. Soc. America, 22, 357, (1932); Ref. 2: Wick F. G., Troupp Ch. G., J. Opt. Soc. America, 25, 57, (1935). Card 2/3





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#### "APPROVED FOR RELEASE: 04/03/2001 CIA-

CIA-RDP86-00513R001756620017-6

Ural'skoye soveshchaniye po spektroskopii. 3d, Sverdlovsk, 1960.

Materialy (Materials of the Third Ural Conference on Spectroscopy) Sverdlovsk, Metallurgizdat, 1962. 197 p. Errata slip inserted. 3000 copies printed.

Sponsoring Agencies: Institut fiziki metallov Akademii nauk SSSR. Komissiya po spektroskopii; and Ural'skly dom tekhniki VSMTO.

Eds. (Title page): G. P. Skornyakov, A. B. Shayevich, and S. G. Bogomolov; Ed.: Gennadiy Pavlovich Skornyakov; Ed. of Publishing House: M. L. Kryzhova; Tech. Ed.: N. T. Mal'kova.

PURPOSE: The book, a collection of articles, is intended for staff members of spectral analysis laboratories in industry and scientific research organizations, as well as for students of related disciplines and for technologists utilizing analytical results.

COVERAGE: The collection presents theoretical and practical problems of the application of atomic and molecular spectral analysis in controlling the chemical composition of various materials in ferrous and nonferrous metallurgy, geology, chemical industry, and medicine. The authors express their thanks to G. V. Chentsova for help in preparing the materials for the press.

References follow the individual articles.

Materials of the Third Ural Conference (Cont.)	sov/6181
Genkin, A. M., and S. G. Bogomolov. Explanation of the mechanism of interaction between proteins and glycogen by optical methods	183
Grebenshchikova, M. P., K. V. Mukhorina, and S. G. Bogomolov Absorption spectra of potato juice treated with diethanol amine salt of hydrazide maleic acid	
Trofimov. A. K. Spectral-luminescence method for investigating crystallochemical transformations in solid phases	190
Trofimov, A. K. Quantitative determination of gadolinium traces in fluorite, metallic thorium, and beryllium by luminescence spectra	192
Florinskaya, V. A., and R. S. Pechenkina. Application of infrared spectroscopy to the study of silicate structure	194
Card 14/15	

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TROFINCY, A.M.

AUTHOR TITLE

PERIODICAL ABSTRACT GREBENSHCHIKOVA, V.I., TROFIMOV, A.M.

The All-Union Congress on Radiochemistry.

(Vsesoyusnoye soveshchaniye po radiokhimii... Russian)

(Vsesoyusnoye soveshchaniye po radiokhimii... Russian)

Atomnaya Energiya 1957, Vol 2, Nr 6, pp 562-563 (USSR).

This congress, which took place at Leningrad, was attended by about 600 scientists from various cities of the country. The 50 lectures delivered on this Congress dealt with the main problems of theoretical radiochemistry and the chemistry of some radioactive elements. Also the form of the existence of small quantities of radioactive substances in solutions and solids, as well as their behavior on the occasion of precipitation with carriers, the laws of the distribution between two non-mixing phases, the chemistry of technetium, promethium, and the transplutonium elements (americium, ourium, berkelium, californium) um elements (americium, ourium, berkelium, californium) etc. was dealt with. The central problem of radioactive consists in the investigation of the state of radioactive elements in diluted solutions and of their behavior on the occasion of precipitation with crystalline deposits.

CARD 1/3

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The All-Union Congress on Radiochemistry.

I.E.STARIK and his collaborators established the fact of the existence of true colloids in substances which are in solution in extreme dilutions. By this the production of truly colloidal solutions of radioactive substances, which had been denied for a long time, is confirmed.

By combining different methods of investigation for colloidal solutions it is possible, univocally to determine the portion of the matter existing in the solution in the iron state as well as in form of true colloids and pseudocolloids. The Congress also discussed several problems concetted with the application of adsorption processes in chemical practice. The results of theoretical and experimental investigations concerning the following problems were dealt with:

Theory of ion exchange, chromatographical separation of theory of ion exchange, chromatographical separation of the state rare earths and transuranium, determination of the state of radioactive elements in a solution by their adsorption of on glass, and ion-exchange-resins, selective adsorption of some radioactive elements on ion-exchange resins, silicatesome radioactive elements on ion-exchange resins, silicategel, and on other porous adsorbents. The Congress arranged

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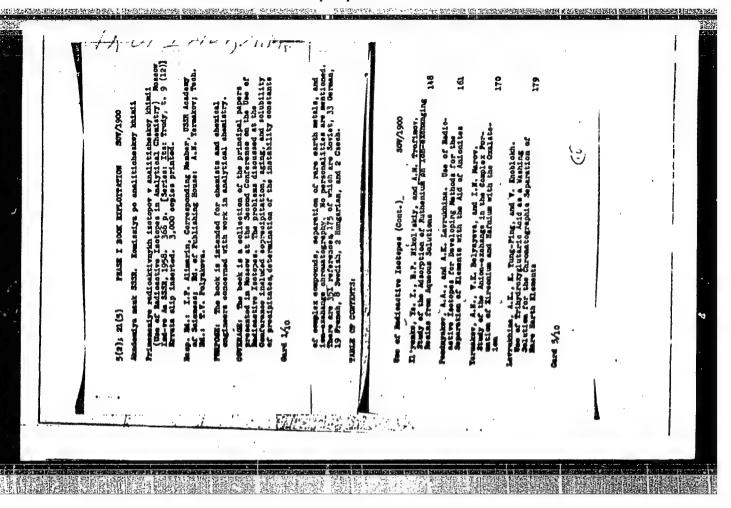
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The All-Union Congress on Radiochemistry.

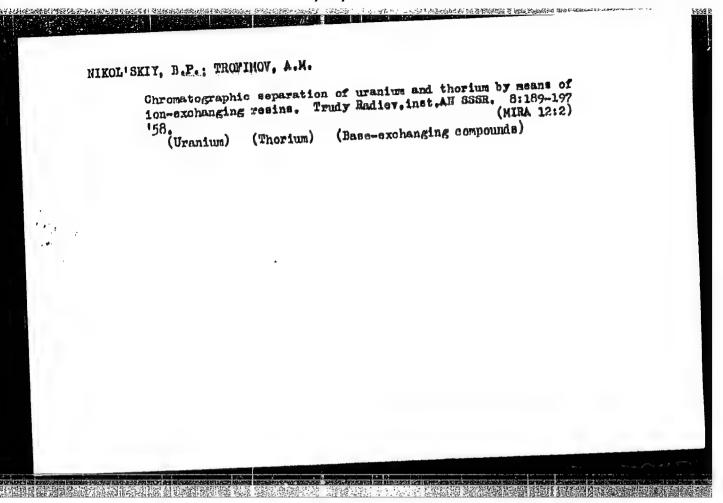
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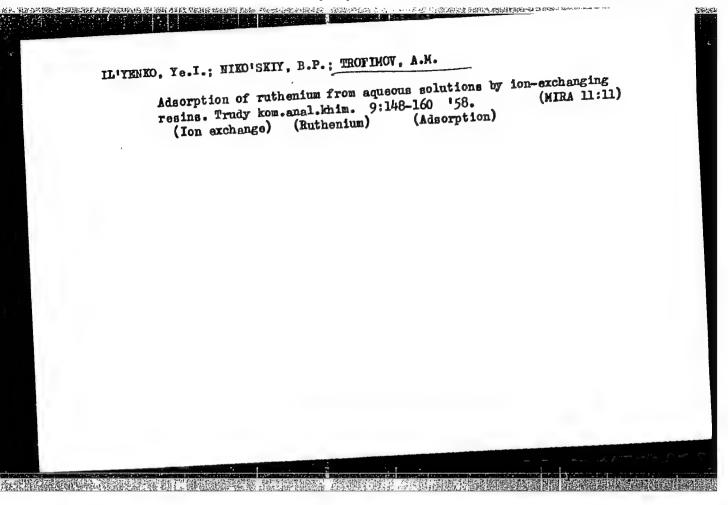
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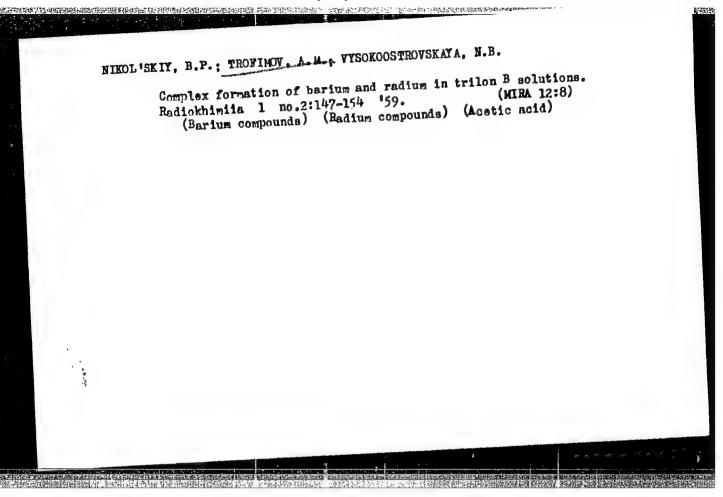
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APPROVED FOR RELEASE: 04/03/2001 CIA-RDP86-00513R001756620017-6"







NIKOL'SKIY, B.P.; TROFIMOV, A.M.; VYSOKOOSTROVSKAYA, N.B.

Reaction of radium and barium with nitrilotriacetic acid in aqueous solutions. Radiokhimiia 1 no.2:155-161 '59.

(MIRA 12:8)

(Radium) (Barium) (Acetic acid)

NIKOL'SKIY, B.P.; TROFIMOY, A.M.: PANFILOVA, G.G.

Admorption of zirconium and niobium by silica gel. Radiokhimiia (MIRA 12:10)

1 no.3:283-289 '59. (Zirconium) (Miobium) (Silica)

TROFIMOV, A.M.; STEPANOVA, L.N.

Change in the magnitude of the charge of zirconium ions in a nitric acid solution, as determined by means of ion exchange resins. Radio--khimiia 1 no.4:403-407 '59. (MIRA 13:1)
(Zirconium--Isotopes)

5(4)

SOV/78-4-4-24/44

AUTHORS:

Nikol'skiy, B. P., Trefimov, A. M., Vysokoostrovskaya, N. B.

TITLE:

Investigation of the Behavior of Potassium Ions in Solutions

of Ethylenediamine Tetraacetic Acid by the Ion Exchange

id a Potassium Glass Electrode (Issiedovanilya povedeniya longv kaliya v rastvorakh etilendiamintetrauksusnoy kisloty s pomoshch'yu ionnogo obmena i kaliyevogo steklyannogo

elektroda)

PERIODICAL:

Zhurnal neorganicheskoy khimii, 1959, Vol 4, Nr 4, pp 857-861

(USSR)

ABSTRACT:

The authors investigated the behavior of potassium ions in solutions of ethylenediamine tetraacetic acid (EDTA) by the method of ion exchange by means of the radioactive indicator

 $\kappa^{42}$  and by the potentiometric method by means of a potassium glass electrode. The interaction of potassium with EDTA was investigated by means of the cation exchanger KU-2 in the Na form and by means of the anion exchanger AV-17 in the Cl form. The results are listed in table 1. It was found that

Card 1/2

potassium ions within the pH range 6-11 form no complex with

SOV/78-4-4-24/44

4 (1) 4 MATERIAL PROPERTY AND A STREET PARTY OF THE STREET PARTY O

Investigation of the Behavior of Potassium Tons in Solutions of Ethylenediamine Tetraacetic Acid by the Ion Exchange Method and a Potassium Glass

Electrode

EDTA. In order to check this statement, the authors made experiments concerning the adsorption of potassium on the anyion exchangers AV-17 and Dowex-1 from solutions without sodium ions and with an EDTA concentration of 0.25 m at pH 7.6-11. The results are given in table 2. The potentiometric investigations indicated that with increasing EDTA concentration no complex is formed in the solution since the electromotive force of the galvanic cell remains constant. The results of the potentiometric investigations are given in table 3. There are 3 tables and 8 references, 6 of which are Soviet.

ASSOCIATION:

Radiyevyy institut AN SSSR im. V. G. Khlopina (Radium Institute

of the AS USSR imeni V. G. Khlopin)

SUBMITTED:

June 21, 1958

Card 2/2

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TROFINGY, A.M.

175

PHASE I BOOK EXPLOITATION SOV/5410

Tashkentskaya konferentsiya po mirnomu ispol'zovaniyu atomnoy energii, Tashkent, 1959.

Trudy (Transactions of the Tashkent Conference on the Peaceful Uses of Atomic Energy) v. 2. Tashkent, Izd-vo AN UzSSR, 1960. 449 p. Errata slip inserted. 1,500 copies printed.

Sponsoring Agency: Akademiya nauk Uzbekskoy SSR.

Responsible Ed.: S. V. Starodubtsev, Academician, Academy of Sciences Uzbek SSR. Editorial Board: A. A. Abdullayev, Candidata of Physics and Mathematics; D. M. Abdurasulov, Doctor of Medical Sciences; U. A. Arifov, Academician, Academy of Sciences Uzbek SSR; A. A. Borodulina, Candidate of Biological Sciences; V. N. Ivashev; G. S. Ikramova; A. Ye. Kiv; Ye. M. Lobanov, Candidate of Physics and Mathematics; A. I. Mikolayev, Candidate of Medical Sciences; D. Mishanov, Candidate of Chemical Sciences; A. S. Sadykov, Corresponding Member, Academy of Sciences USSR, Academician, Academy of Sciences Uzbek SSR; Yu. N. Talanin,

Card 1/20

Transactions of the Tashkent (Cont.)

Candidate of Physics and Mathematics; Ya. Kh. Turakulov, Doctor of Piological Sciences. Ed.: R. I. Khamidov; Tech. Zd.: A. G. Babakhanova.

PURIOSE: The publication is intended for scientific morkers and specialists caployed in enterprises where radicactive isotopes and nuclear radiation are used for research in chemical, geological, and technological fields.

COVERAGE: This collection of 133 articles represents the second volume of the Transactions of the Tashkent Conference on the Faureful Uses of Atomic Energy. The individual articles deal. With a wide range of problems in the field of nuclear radiation, including; production and chemical analysis of radicactive isotopes; investigation of the kinetics of chemical reactions by means of isotopes; application of spectral analysis for the manufacturing of radicactive preparation; radicactive methods for determining the content of elements in the rocks; and an analysis of methods for obtaining pure substances. Certain

Card 2/20

Transactions of the Tashkent (Cont.)

SOV/5410

instruments used, such as automatic regulators, flowactors, level gauges, and high-sensitivity gamma-relays, are described. No personalities are mentioned. References follow individual articles.

TADLE OF CONTENTS:

RADIOACTIVE ISOTOPES AND NUCLEAR RADIATION
IN ENGINEERING AND GEOLOGY

Lobanov, Yo. M. [Institut yaderney fiziki UzSSR - Institute of Nuclear Physics AS USSSR]. Application of Radioactive Isotopes and Nuclear Radiation in Uzbekistan

Taksar, I. M., and V. A. Yanushkovskiy [Institut fiziki AN Latv SSR - Institute of Physics AS Latvian SSR]. Problems of the Typification of Automatic-Control Apparatus Based on the Use of Radioactive Isotopes

Card 3/20

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Granzactions of the Tashkent (Cont.)	30V/5410	
Radium Institute imeni V.G. Kilopin AS USSR]. quantities of Radioactive Elements in Solutions	State of the Mero-	3
Trofimov, A. M., and L. H. Stepanova [Radium I 7.6. Khlopin A3 USSR]. Determination of the Ha Charge of Complex Ions of Radioactive Elements change Hathod	antitude of a	
Yerofeyev, B. V., and V. A. Protashchik [Insticted Organic Chemistry AS BelSSR]. Application of the Surface Size of Schid Substances	n of Badioactive	3
I-win, V. I., and V. V. Bochkarev [Ministry of Obtaining Radioactive Isotopes in the Reactors Ehreshhold, Consecutive, and Secondary Ruclear R	by Coons of	8
Bukharov, I. N. [Ministry of Health USSR]. Pe Lientification and Analysis of the Tagged Organ	culiarities in ic Compounds 37	2
Card 17/20		
		1

S/054/60/000/004/007/015 B004/B056

AUTHORS:

Trofimov, A. M., Stepanova, L. N.

TITLE:

Investigation of the Exchange of Ions of Different Valences on Swelling Ion Exchangers and Application of the Rules Found for the Determination of the Ion Charge in the Solution

PERIODICAL:

Vestnik Leningradskogo universiteta. Seriya fiziki i khimii, 1960, No. 4, pp. 70-76

TEXT: Proceeding from B. P. Nikol'skiy's theory of ion exchange, the ion exchange in highly swelling exchange resins has been studied by radiochemical methods at the Radiyevyy Institut AN SSSR (Radium Institute of the AS USSR). It was experimentally established that the different swelling capacity greatly affects the selective adsorption in the exchange of ions of different valency. This is indicative of a different concentration of adsorbed ions in the resin. The ion exchange of Ky -2 (KU-2)-type and MC $\Phi$  (MSF)-type resins was investigated by means of Ce 44, Ra 226, and Cs 134. The following equation was derived for calculating the ion charge z:  $z = \left[\log(\alpha^{I}/\alpha^{II}) + \log(v^{II}/v^{I})\right] / \left[\log(g^{I}/g^{II}) + \log(v^{II}/v^{I})\right]$ . Here, Card 1/2

Investigation of the Exchange of Ions of S/054/60/000/004/007/015 Different Valences on Swelling Ion Exchangers B004/B056 and Application of the Rules Found for the Determination of the Ion Charge in the Solution

α<sup>I</sup>, α<sup>II</sup> denote the distribution coefficients which were determined experimentally in resins with different specific volumes VI, VII, and different specific capacities g<sup>I</sup>, g<sup>II</sup>. This method of different ion concentrations in the resin phase was used to determine the charge of zirconium ions (Ref. 9) and, together with A. A. Grinberg, to determine the charge of ruthenium complexes (Ref. 10). G. V. Samsonov and A. B. Pashkov are mentioned. There are 2 tables and 10 references: 5 Soviet, 3 US, 1 British, and

Card 2/2

\$/186/60/002/001/013/022 A057/A129

Grinberg, A.A.; Trofimov, A.M.; Stepanova, L.N.

TITLE:

Determination of the charge of polynuclear complex ruthenium ions by

the ion-exchange method

PERIODICAL: Radiokhimiya, v. 2, no. 1, 1960, 78 - 82

The present investigation was carried out after a visit of one of TEXT: the present authors in the-laboratory of J.M. Fletcher in Harwell (England) in connection with some new data (reported by Fletcher et al. at the International Conference on Coordination Chemistry, London, May 6, 1959, under the title: binuclear chloro and other polynuclear complexes of ruthenium) concerning ruthenium complexes. In the discussion the investigators stated the importance of direct determination of the charge of the red polynuclear ruthenium cation, for which the British chemists assumed a charge of +6. Definite solution of this question was of interest apart from the verification of data obtained by Fletcher et al., because complex anions with charges greater than four are rare. F.M. Jaeger and P. Koets [Ref. 3: Z] anorg. Ch., 170, 347 (1928)] reported about nine-valent cations, but their existence is at present in question [J.C. Bailar, Ref. 4: Chem-

Card 1/5

Determination of the charge of polynuclear...

S/186/60/002/001/013/022 A057/A129

istry of the Coordination Compounds, 65, N.Y. (1956)]. Hence it was important to discover a method to determine the charge of highly-charged cations. Thus the present authors investigated the applicability of the recently published ion-exchange method [A.V. Trofimov and L.N. Stepanova, Ref. 2: Radiokhimiya, 1, 4, 403 (1959)] to the determination of the charge of the red polynuclear ruthenium cation. In further investigations this method will be applied to check data obtained by Jaeger and Koets. In the present experiments a sample of the ruthenium complex synthesized by Fletcher et al. was used. The principle of the ion-exchange method consists in the determination of the distribution coefficient  $\alpha$  of radioisotopes on two ion-exchange resins with different swelling capacities. According to the rules of ion-exchange:  $\lg \frac{e^I}{e^{II}} = \frac{s_1}{s_2} \lg \frac{G^I}{G^{II}} + \frac{s_1-s_2}{s_2} \lg \frac{V^{II}}{V^I}$  (1)

I and II refer to the resins with two swelling capacities,  $z_1$  - effective charge of the investigated ions;  $z_2$  - charge of the exchanged ions,  $C^I$  and  $C^{II}$  equivalent exchange capacity of the resins (per 1 g of dry resin),  $V^I$  and  $V^{II}$  - specific volumes of swellen resins under the conditions of the distribution coefficient determination. In the exchange of mono-valent ions (H<sup>†</sup>, Na<sup>†</sup> etc.), the charge can be calculated by:  $\frac{1}{2} e^{a^I} + \frac{1}{2} e^{V^{II}}$ 

 $s = \frac{\lg_{a^{11}} + \lg \frac{1}{v^{1}}}{\lg_{a^{11}}^{G^{1}} + \lg \frac{v^{11}}{v^{1}}} (2)$ 

and if the equivalent exchange capacities of the result is same:  $\frac{\lg \frac{a!}{a!!}}{\lg \frac{v!!}{v!}} + 1 (3)$ 

Card 2/5

Determination of the charge of polynuclear....

S/186/60/002/001/013/022 A057/A129

The experimental determination of a as well as of the specific volumes of the swollen ion-exchange resin must be carried out under the same conditions. The investigated element must be ions. The ion-exchange must be strictly reversible and the complexes must be stable. According to Ye.I. Il yenko, B.P. Nikol'skiy and A.M. Trofimov [Ref. 5: Tr. komissii po analiticheskoy khimii (Proceedings of the commission for analytical chemistry), Izd. AN SSSR (Ed. AS USSR), 9 (12), 148 (1958)] reversibility is not always maintained in exchange of ruthenium complexes. The present authors demonstrated in corresponding experiments that by adding HNO2 solution the red complex changes into a yellow complex, thus exchange using H+ Ions cannot be carried out. It was observed that in NaNO2 solutions the complex is stable, and is strongly adsorbed on sulfonated KY-2 (KU-2) cation exchange resin. About 50% of the red complex is adsorbed from 3.5 N NaNO3 solution. Solutions containing between 0.5 and 5 mg/l ruthenium obey Beer's law with an absorption maximum at 460 mm. Thus the present experiments were carried out with concentrations of 1.5 mg Ru/1, reversibility was tested and or was determined as ~3,400. Two samples of the resin (containing 2% or 12% divinylbenzene) were soaked in 3.5 N NaNO3 solution and the specific volumes were determined picnometrically with octane resin with 2% divinylbenzene 1.83 ± 0.01 ml/g; with 12% divinylbenzene 1.37 ± 0.01 ml/g. The swelling capacity is doubled in water.

Card 3/5

Determination of the charge of polynuclear....

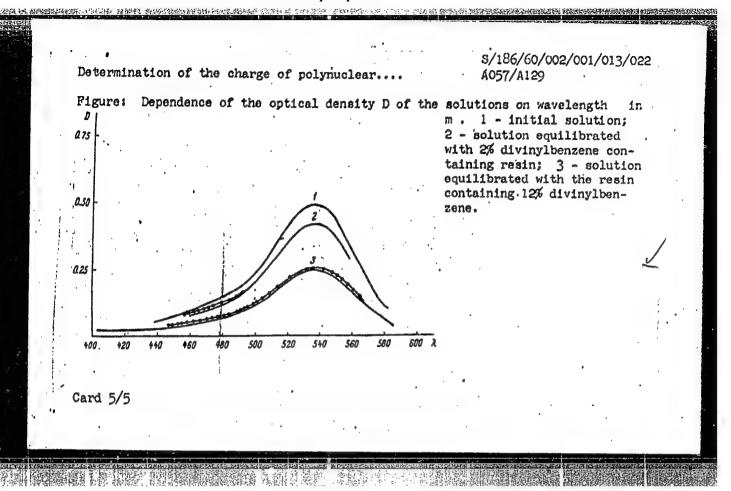
等价值的证明的19-45cm(1-13-15cm和股份ANCE的图象是否的通知和19-50-50 通过多数的第三人称形式。这些形式是否是否是多

S/186/60/002/001/013/022 A057/A129

Since the exchange capacities for both resins are 4.83 - 4.85 mg equiv/g the calculations were done according to equation (3). The concentration of ruthenium in the initial and in equilibrated solutions was determined with a recording  $C\Phi$ -2M (SF-2M) spectrophotometer and  $\Phi$ 3K-2M (FEK-2M) photoelectrocolorimeter using green filters. From the obtained results (see Fig.) the charge of the complex was calculated with z = 5.9. Thus data presented by Fletcher et al. are confirmed; on the other hand it is demonstrated that the present method can be used for determinations of the charge of polynuclear complexes. There are: 1 figure and 5 references: 2 Soviet-bloc and 3 non-Soviet-bloc.

SUBMITTED: November 13, 1959

Card 4/5



Study of the exchange of ions of various valences on swelling ion exchangers, and application of the mechanisms discovered to

TROFINOV, A.M.; STEFANOVA, L.N.

the determination of the ionic charge in solution [with summary in English]. Vest. LOW 15 no.22:70-76 '60. (MIRA 13:11) (Ion exchange)

S/076/60/034/008/029/039/XX B015/B063

26.1610 AUTHORS:

Trofimov, A. M. and Stepanova, L. N.

TITLE:

Radiochemical Study of Ion Exchange on Swollen Ion Exchangers

PERIODICAL:

Zhurnal fizicheskoy khimii, 1960, Vol. 34, No. 8,

pp. 1837 - 1842

TEXT: Contrary to K. K. Gedroyts (Ref.1), B. P. Nikol'skiy (Ref.2), and Ye. N. Gapon (Ref.3) who discussed ion exchange with standard ion exchangers, the present authors discuss the behavior of swollen ion exchangers. This subject has also been discussed by Gregor (Ref.4), changers. This subject has also been discussed by Gregor (Ref.4), changers. This subject has also been discussed by Gregor (Ref.4), changers. V. Samsonov (Ref.5), and Griessbach (Ref.6). Experiments have shown that the difference in the swelling capacity of ion exchangers has a particularly strong effect on the selectivity of exchange of ions of different valencies. The selective adsorption of ions of higher valency sharply increases with a decrease of the swelling capacity of the exchanger. This is ascribed to the varying ion concentration in the solid changer. This is ascribed to the varying ion concentration in the solid changer of ion exchangers with different swelling capacity. The rule of this phase of ion exchangers with different swelling capacity. The rule of this phenomenon was theoretically and experimentally studied by the radio-

863,55

Radiochemical Study of Ion Exchange on Swollen Ion Exchangers

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\$/076/60/034/008/029/039/XX B015/B063

chemical method. The selectivity of adsorption of a radioactive element on two different ion exchangers may be determined from the ratio between the distribution coefficients  $\alpha$ ;  $\alpha_1^{I/\alpha_1^{II}} = (g^{I/g^{II}})^{z_1/z_2} \cdot (v^{II/v^{I}})^{z_1-z_2/z_2} \cdot (f_2^{I/f^{II}})^{z_1/z_2} \cdot (f_1^{I/f^{II}})$  (8), where I and II refer to the two exchangers; g is the absorbed quantity of ions per weight unit of the exchanger; V is the specific volume of the swollen exchanger;  $z_1$  and  $z_2$  denote the ion valency; and  $f_1$  and  $f_2$  are their activity coefficients. g and V may be easily determined by way of experiment. The activity coefficients can be represented by the function  $\phi(f) = (f_2^{I/f^{II}})^{z_1/z_2} \cdot f_1^{II/f^{I}}(9)$ . The experiments were performed with MC $\Phi$  (MSF) and Ky-2 (KU-2) exchangers which had been made available by a solutions was studied. The measurements indicate that the swelling capacity of an exchanger greatly affects the distribution of ions of different valencies among exchanger and solution. Using the equation

Card 2/3

Radiochemical Study of Ion Exchange on Swollen Ion Exchangers

S/076/60/034/008/029/039/XX B015/B063

 $(z_1-z_2)/z_2/g^2/z_1/z_2$  = const. it is possible to determine the valency of a radioactive element in a solution by using two exchangers with equal is thanked for a discussion. Polyanskiy is mentioned. There are 3 tables and 6 references: 4 Soviet, 1 US, and 1 German.

ASSOCIATION: Akademiya nauk SSSR Radiyevyy institut im. V. G. Khlopina (Academy of Sciences USSR, Radium Institute imeni

SUBMITTED: December 7, 1958

Card 3/3

STUPISHIN, A.V., prof.; BABANOV, Yu.V., ml. nauchn. sotr.;

GUSEVA, A.A., ml. nauchn. sotr.; DUGLAV, V.A., dots.;

ZAKHAROV, A.S., dots.; KOSTINA, N.M., assistent; LAVROV,

D.D., dots.; LAPTEVA, N.N., assistent; ROMANOV, D.F., ml.

nauchn. sotr.; SIROTKINA, M.M., aspirant; SMIKNOVA, T.A..

ml. nauchn. sotr.; TORSHYEV, N.P., st. prepod.; TAYSIN.

A.S., st. prepod.; TROFIMOV, A.M., assistent; KHARITONYCHEV,

A.T., prepod.; STUPISHIN, A.V., red.; KHABIBULLOV, R.K.,

red.

[Establishing physicogeographical regions in the middle Volga Valley] Fiziko-geograficheskoe raionirovanie Srednego Povolz'ia. Kazan', Izd-vo Kazanskogo univ., 1964. 196 p. (MIRA 18:12)

TROFIMOV, A.M.; KAZANKIN, Yu.N.

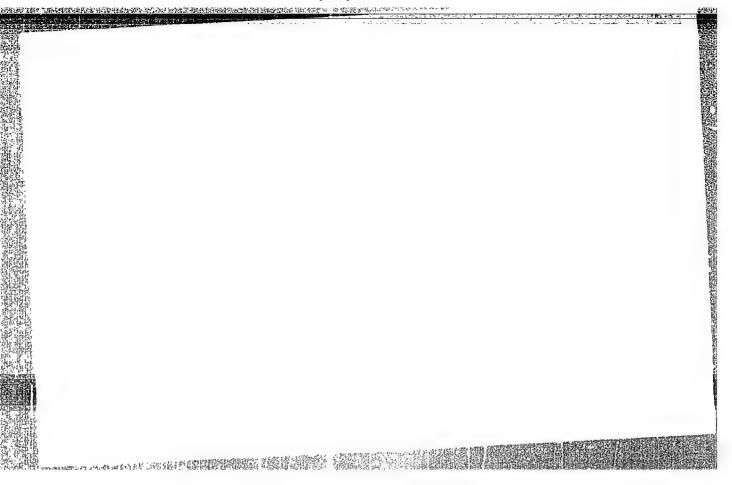
Clathrate compounds of p-cresol with noble gases. Part 1: p-Cresol compound with xenon. Radiokhimiie 7 no.3:288-292 '65. (MIRA 18:7)

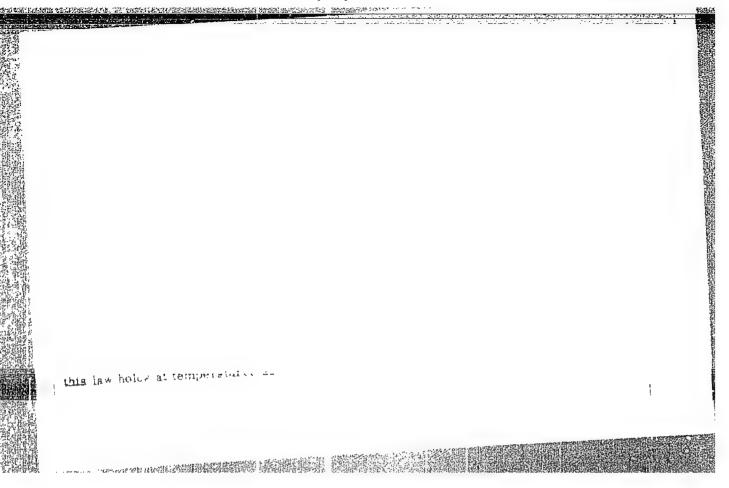
TROFINOV, A.M., PANKOV, A.M.

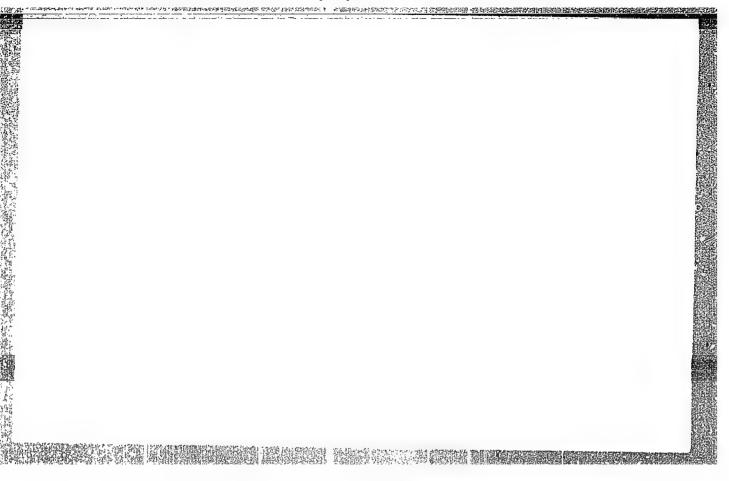
Effect of a macrocomponent gas on the distribution of Kr85 and Xe<sup>133</sup>
between the gaseous phase and hydrocarbon sorbent. Radiokhimita 7 no.3:
(MIRA 18:7)
293-298 '65.

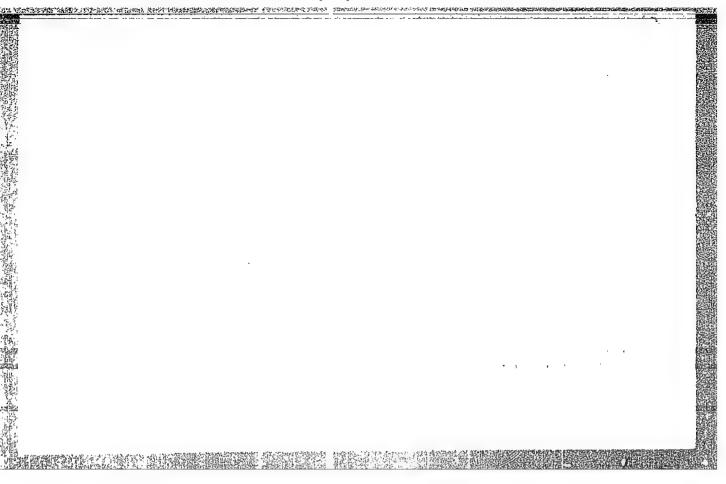
TROPIMOV, A.M.; FANKOV, A.M.; KOVARSKIY, A.P.

Preparation and study of the stability of polymethyl methacrylate cryptonates. Radiokhimiia 7 no.3:359-361 '65. (MIRA 18:7)











TROFIMOV, Aleksey Mikhaylovich; VLASOV, A.G., inzh., retsenzent;

KHAYMOVICH, Ye.M., doktor tekhn. nauk, prof., red.;

NIKIFOROVA, R.A., inzh., red.; GORNOSTAYFOL'SKAYA, M.S.,

tekhn. red.

[Album of machine-tool designs] Al'bom skhem metallorezhushchikh stankov. Moskva, Mashig. Pt.1. [Lathes, drilling and boring machiner,] Tokarnye, sveril'nye i rastochnye stanki. 1961.

50 diagrams. [Description] Opisanie. 137 p. (MIRA 15:5) (Lathes) (Drilling and boring machinery)

TROFIMOV, Aleksey Mikhaylovich; STOLYAR, N.M., inzh., retsenzent;

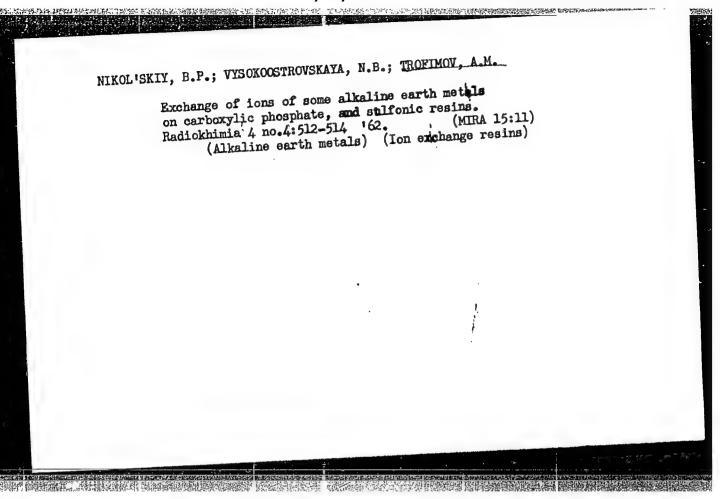
KHAYMOVICH, Ye.M., doktor tekhn. nauk, prof., red.;

NIKIFOROVA, R.A., inzh., red.; GORNOSTAYPOL'SKAYA, M.S.,

tekhn. red.

[Album of diagrams of metal-cutting machines]Al'bom skhem metalloreshushchikh stankov. Moskva, Mashgiz. Pt.2.[Milling, thread-cutting, planing, broach-grinding, dressing, gear-cutting machines and machine-assemblies]Frezernye, rez'bonareznye, strogal'nye, protiazhnye shlifoval'nye, zatochnye, zuboobrabatyvaiushchie, agregatnye stanki. 1962. 69 p. \_\_\_ [Description]Opisanie. 252 p. (MIRA 16:1)

(Gutting machines)



TF	Improve the quality of switches. Put' i put.khoz. 5 no.6:25  (MIRA 14:8 Je '61.  1. Dorozhnyy master, st.Dno, Oktyabr'skoy dorogi.  (RailroadsSwitches)

Characteristics of the development of slopes. Ter. vy.: unhabed 22v.; geol. i razv. 6 no.5:35-37 My 165. (Minh 18:10)
l. Kazanskiy gosudarstvennyy universitet imeni Ul'yanova

STIPLING, A.V.; TORSUYEV, N. P.; TROFLMOV, A.M.

A new Marst hole. Izv. Vses. Geog. ob-vs 97 rc.5:461-463
(MIRA 18:11)

S.O. 165.

YEFREMOV, I.S., doktor tekhn. nauk; REKITAR, R.A., inzh.;

ROZENBERG, S.V., kand. ekon. nauk; BLATNOV, M.D., kand.

tekhn. nauk; VIL'KONETSKIY, M.S., inzh.; TCMILIN, A.I., inzh.;

POPELYASH, V.N., inzh.; ZAGAYNOV, N.A., kand. tekhn. nauk;

FINKEL'SHTEYN, B.S., inzh.; MARINOV, I.A., inzh.; ISTRATOV, V.P.,

inzh.; MARGOLIN, I.S., inzh.; ENGEL'S, G.G., inzh.; ANTONOV,

V.A., inzh.; SOKOLOV, V.D., inzh.; KLESHCHINSKIY, B.K., inzh.;

IL'INSKIY, A.I., retsenzent; PAPKOV, N.G., retsenzent; STIRNOV,

G.M., retsenzent; SHPOIYANSKIY, M.N., otv. red. toma; VOLOCHNEV,

V.N., red.; TROFIMOV, A.N., red.; RACHEVSKAYA, M.I., red. izd-va;

LEIYUKHIN, A.A., tekhn. red.

[Technical manual on city electric transportation in three volumes] Tekhnicheskii spravochnik po gorodskomu elektrotransportu v trekh tomakh. Redkollegiia: V.N.Volochnev, A.N. Trofimov, M.N.Shpolianskii. Moskva, Izd-vo M-va kommun. khoz. RSFSR. Vol.1. [City electric transportation (general part)] Gorodskoi elektricheskii transport (obshchaia chast'). Otv. red. toma M.N.Shpolianskii. 1961. 726 p. (MIRA 15:4) (Streetcars)

## TROFIMOV, A.M.

Current distribution on the cethode during electrodeposition of metals in the ultrasonic field. Elektrokhimiia 1 no.9:1150-(MIRA 18:10)

l. Bashkirskiy gosudarstvennyy universitet imeni 40-letiya Oktyabrya.

Onlocalisation of the correct distribution on a catacate by the mathod of abreight lines. Flekbockhimins is on established in 165.

1. Bashkirskiy gosudarstvennyy universited intri (Onlettya Oktyabrya.

IVIN, Kirill Vladimirovich; TROFIMOV Aleksanir Hassrovich, ENGEL'S,
Georgiy Georgiyevich; KRIDV, S.K., redaktor; CROCHEVA, M.A.,
redaktor izdatel'stva; FOMBERG, P.I., tekhnicheskiy redaktor

[Trolley buse collectors] Tokos\*em trolleibusa. Noskva, Izd-vo
Ministerstva kommunal'nogo khoziaistva RSFSR, 1956. 191 p.

(Trolley buses)

(Electric current collectors)

1087, 2808, 2208, 2607

\$/194/61/000/001/025/038 D216/D304

1 1800

AUTHOR:

Trofimov, A.N.

TITLE:

Distribution of metal at the cathode surface during

electro deposition in an ultra-sonic field

PERIODICAL:

Referativnyy zhurnal. Avtomatika i radioelektronika, no. 1, 1961, 18, abstract 1 El53 (V Sb. Primeneniye

ul'traakust. k issled. veshchestva, no. 10, M.,

1960, 103-107)

The results are given for the experimental analysis of metal distribution at the cathode surface during electro-deposition of copper, zinc and nickel in an ultrasonic field and also on the influence of the ultrasonics on the cathode polarization and on metal output as a function of current. It is shown that changes in the dispersing ability of baths correspond exactly to changes in the slope of the polarization curve and metal output under the influence of an ultrasonic frequency of 30 Kc/s and at field inten-

Card 1/2

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Distribution of metal...

sities of 0.15 and 0.3 W/cm<sup>2</sup>. It is explained that with the increase of ultrasonic intensity, the irregularity of the copper distribution at the cathode surface increases. It has been noticed that a low frequency ultrasound has very much the same effect on dispersing properties of the bath as stirring. 3 references.

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Card 2/2

(O)8 AUTHORS: SOV/105-5y-12-21/23 Chilikin, M. G., Tikhomirov, S. S., Trofimov, A. N., Ivanov, I.T., Rozenfel'd, Y. Ye., Minov, D. K., Medel', V. B.

TITLE

Professor I. S. Yefremov, On His 50th Birthday

PERIODICAL:

Elektrichestvo, 1959, Nr 12, p 83 (USSR)

ABSTRACT:

Ivan Semenovich Yefremov was born in July 1909. in 1935 he graduated from the fakul tet elektrifikatsii (Department of Electrification) of the Moskovskiy elektromekhanicheskiy institut inzhenerov sheleznodorozhnogo transporta (Moscow Electromechanical Institute for Railroad Engineers). He is working since then at the Trolley Administration of Moscow, where he became plant manager, after being foreman and chief engineer. He takes part in the scientific work of the research laboratory of the gorodskoy elektricheskiy transport Akademii kommunal'nogo khozyayatva (Municipal Electrical Transportation of the Academy of Communal Economy). In 1946 he graduated as Candidate of Technical Sciences, in 1949 he was elected the chief of the kafedra elektricheskoy tyagi i podvizhnogo sostaya Moskovskogo avtodoroznnogo instituta (Chair of Electrical Traction and Vehicles of the Moscow Institute of Highways) ~-

Card 1/2

Professor I. S. Yefremov. On His 50th Birthday

sov/105-59-12-21/23

In March 1956 he became head of the kafedra elektricheskogo transports of the Moskovskiy energeticheskiy institut (Chair of Electrical Transportation of the Moscow Institute of Power Engineering). He still holds this position. In April 1959 he Engineering). He still holds this position. In April 1959 he became dean of the fakulitet elektrifikatsii promyshlennosti transporta MEI (Department of Electrification of the Industry and Transportation at the Moscow Institute of Power Engineerand Industry National Sciences and Industry National Sciences Industry National Sciences Industry National Sciences Industry National Sciences Industr

Card 2/2

IVIN, Kirill Vladimirovich; TROFIMOV, Aleksandr Nazarovich;
ENCEL'S, Georgiy Georgiyevich

[Pantographs of municipal surface transportation] Tokos\*em gorodskogo nazemnogo transporta. Moskva, Stroiizdat, 1965. 261 p.

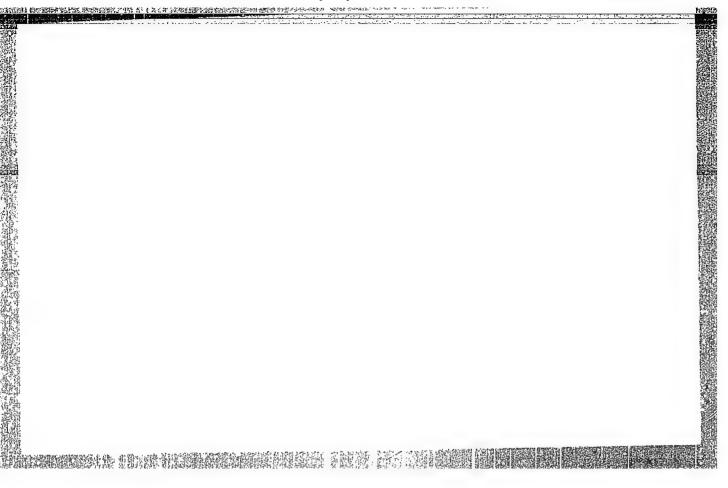
(MIRA 18:7)

TROFIMOV, A.N.; GALUSHKO, A.P.

Outhodic polarization in the electrodeposition of metals in the Ultrasonic field. Elektrokhimia 1 no.81985-988 Ag 165. (MIRA 18:9)

1. Bashkirskiy gosudarstvennyy universitet imeni 40-letiya Oktyabrya.





TROFIMON, A.N.

PHASE I BOOK EXPLOITATION

SOV/5644

Vserossiyskaya konferentsiya professorov i prepodavateley pedagogicheszikh institutov

Primenentye ul' trankustiki k issledovaniyu veshchestva. vyp. 10. (Utilization of Ultrasonics for the Investigation of Materials. no. 10) Moscow, Izd-vo MOPI, 1960. 321 p. 1000 copies printed.

Eds.: V. F. Nozdrev, Professor, and B. B. Kudryavtsev, Professor.

PURPOSE: This book is intended for physicists and engineers interested in ultrasonic engineering.

COVERAGE: The collection of articles reviews present-day research in the application of ultrasound in medicine, chemistry, physics, metallurgy, ceramics, petroleum and mining engineering, defectoscopy, and other fields. No personalities are mentioned. References accompany individual articles.

Card 140

Utilization of Ultrasonics (Cont.)	SOV/5644
Kukoz, F. I. [Novocherkasskiy politekhn. i Polytechnical Institute]. Study of the Ei the Electrolytic Oxidation of Chromium Anode	Teet of officer
Trofimov, A. N. [MGPI im. Lenina-Mosco Institute imeni V. I. Lenin]. The Distr Cathode Surface During Electrodeposition Field	IDUCTOR OF INTOVAL OF
Mai'tsev, N. N., and V. I. Dal' [Dneproperoysk Institute of Chemical T Ultrasound to Intensify Absorption	petrovskiy KhTI - echnology). Using
Mal'tsev, N. N. [Dnepropetrovsk Institute nology]. Study of the Precipitation of C Circulating Waters of a Coal-Enriching	Mai itchiado
Card $4/10$	

APPROVED FOR RELEASE: 04/03/2001 CIA-RDP86-00513R001756620017-6"

Municipal public transportation systems without conductors. Gor. khoz.

Mosk. 34 no.10:18-20 0 '60. (MIRA 13:10)

1. Nachal'nik Upravleniya passazhirskogo transporta Mosgorispolkoma.

(Moscow--Transit systems)

s/194/62/000/006/130/232 D256/D308

1.1700

AUTHOR:

Electrolytic deposition of metals in ultrasonic Trofimov. A.N.

TITLE:

PERIODICAL:

Referativnyy zhurnal. Avtomatika i radioelektronika, Referativnyy zhurnal. Avtomatika i radioelektronika, no. 6, 1962, abstract 6-5-43 sh (V sb. Primeneniye veshchestva, no. 12, M., 1960, ulitraakust.k issled. fields

In the electrolytic deposition of metals several factors re-TEXT: In the electrolytic deposition of metals several factors feature. The the speed of the process. Each electrolyte has a max. cathode duce the speed of the process. Each electrolyte has a max. cathode of the process. que the speed of the process. Each electrolyte has a max. cathode current density which should not be exceeded. The effect of ultracurrent density which should not be exceeded. The effect of ultra-sound on the process of electrolytic deposition was investigated sound on the process of electrolytic deposition was intensities experimentally using a frequency of 27 kc/s at various intensities experimentally using a frequency of an intensity within the range from 0.02 to 10 W/cm<sup>2</sup>. Ultrasound of an intensity in intensifying from 0.02 to 10 W/cm<sup>2</sup> was found to be most effective in intensifying 0.6 to 1 W/cm<sup>2</sup> was found to be most effective higher than 1 W/cm<sup>2</sup> electrolytic coating: the use of intensities higher electrolytic coating; the use of intensities higher than 1 %/cm² often resulted in a reduced thickness of the denomit due to diens electrolytic coating; the use of intensities higher than 1 W/cm<sup>2</sup> often resulted in a reduced thickness of the deposit due to dispersion. Ultrasound increases current yield and increases the quantity card 1/2

APPROVED FOR RELEASE: 04/03/2001

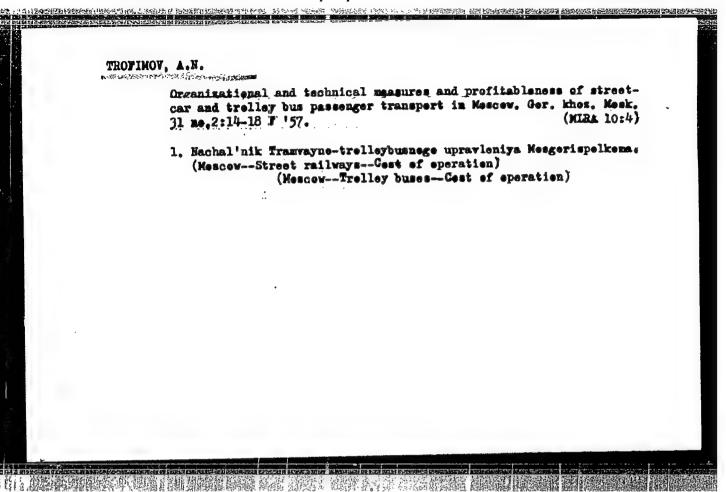
CIA-RDP86-00513R001756620017-6"

Electrolytic deposition of metals ... S/194/62/000/006/130/232 D256/D308

of the deposited metal; the deposit is found to be almost completely free of pores and the process of electrolytic deposition is accelerated. 1 figure, 3 tables, 16 references. [Abstracter's note: Complete translation.]

Card 2/2

<b></b>	Urban passenger transportation. Gor.khoz.Mosk. 35 no.9:35-38 S '61. (MIPA 14:19)
	l. Nachal'nik Upravleniya passazhirskogo transporta Ispolkoma Mossoveta.  (MoscowTransit systems)



# Improvement of the contact network and collectors of Moscow trolley buses. Gor.khoz.Mosk. 24 no.4:23-31 4p '50. (MIRA 7:10) 1. Machal'nik Upravleniya trolleybusnogo transporta Moskvy. (Moscow--Trolley buses) (Trolley buses--Moscow)

TROFINOT, A.N.

Twenty years of motorbus transport service. Gor.khoz.Mosk. 28 no.2:
11-13 F \*54.

1. Machal'nik Upravleniya trolleybusnogo transporta Ispolkoma
Mossoveta. (Moscow-Trolley buses) (Trolley buses-Moscow)

TROFIMOV, A.H., inzhener; ENGEL'S, G.G., inzhener; IVIN, K.V., inzhener

Experience in using elastic contact systems for the trolley bus
lines in Budapest. Gor.khoz.29 no.9:29-33 S'55. (MLRA 8:12)

(Budapest--Trolley buses)

TROYIMOV, A.H.

Effort to increase the profitableness of the Moscow trolley bus system. Gor. khos. Mosk. 25 no.7:33-34 JI '51. (MLRA 6:11)

1. Hachal'nik Upravleniya trolleybusnogo transporta g. Moskvy. (Moscow--Trolley buses) (Trolley buses--Moscow)

76-32-5-39/47

等可引起的原理 (1850年) 图像学术系统 (1860年) 图像设计 (1860年) 图像设计 (1860年)

AUTHOR:

Trofimov, A. N.

TITLE:

The Distribution of Metal on the Cathode Surface in the Electric Deposition of Copper in the Ultrasonic Field (Raspredeleniye metalla na poverkhnosti katoda pri elektro-

osazhdenii medi v ulitrazvukovom pole)

PERIODICAL:

Zhurnal fizicheskoy khimii, 1958, Vol. 32, Nr 5, pp.1172-1174

(USSR)

ABSTRACT:

In order to clear the existing contradictions in investigations of the above mentioned theme the present work determines the metal distribution on the electrode in copper depositions of pyrophosphoric and sulfuric acid electrolytes in the ultrasonic field. The method of the selective cylindrical cathode was used, a generator with a magnetostriction transformer to the frequency of 30 kilocycles serving as source of ultrasound. The experiments were carried out in three different troughs, the composition of the contents of which is given. From the results can be seen that the homogeneity of the metal distribution is deteriorated by ultrasound, especially in the troughs of pyrophosphoric acid at a mean increase of the

Card 1/2

APPROVED FOR RELEASE: 04/03/2001 CIA-RDP86-00513R001756620017-6"

76-32-5-39/47

The Distribution of Metal on the Cathode Surface in the Electric Deposition of Copper in the Ultrasonic Field

current density and amplification of the ultrasound intensity. Based on the investigation of the polarization curves as well as of the metal yield in dependence on the current it was found that the copper distribution on the cathode is deteriorated in the ultrasonic field, which fact was observed with all electrolytes. Therefore the deteriorating effect on the metal distribution must be taken into account in using ultrasound for the improvement of the quality of the copper depositions and for an increase of the productivity of the trough. Finally the author thanks Professor A. P. Kapustin and Professor A. T. Vagramyan. There are 3 figures and 3 references, 2 of which are Soviet.

ASSOCIATION:

Moskovskiy gosudarstvennyy pedagogicheskiy institut im. V. I. Lenina (Moscow State Pedagogical Institute imeni V. I. Lenin)

SUBMITTED:

July 4, 1957

1. Copper-Electrodeposition 2. Electrolysis-

Card 2/2

Ultrasonic factors

յ6կ5կ \$/137/62/000/003/152/191 A052/A101

/. /200 AUTHOR:

Trofimov. A. N.

TITLE:

Electrodeposition of metals in ultrasonic field

PERIODICAL:

Referativnyy zhurnal, Metallurgiya, no. 3, 1962, 86, abstract 31559 (V sb. "Primeneniye ul'traakust. k issled. veshchestva". Moscow,

no. 12, 1960, 113-119)

TEXT: The effect of ultrasound on the process of electrodeposition of metals was studied, in particular, the effect of ultrasound on the metal distribution on cathodesof a complex form, on the hardness of electrolytic deposits, on the porosity of platings, on the electric conductivity of electrolytes. Depending on geometric parameters of the cathode the effect of ultrasound on the process of electrolysis may either improve or impair the uniformity of the metal distribution on the electrode; the degree to which ultrasound affects the uniformity of platings depends on the nature of polarization. Ultrasound causes a considerable reduction of the hardness of Ni-deposits and has no noticeable effect on the hardness of Cu-deposits under adopted conditions of electrolysis. The decrease of microhardness is connected obviously with the decrease of the amount of H<sub>2</sub>

Card 1/2

Electrodeposition of metals in ultrasonic field

S/137/62/000/003/152/191 A052/A101

included in the Ni-deposit, since metallographic studies have detected only a decrease of the grain size of both Cu- and Ni-deposits. The application of ultrasound in the process of electrolysis causes a considerable decrease of the porosity of the deposit, and applied to the Ni-plating it eliminates completely the pitting formation. The attempt to establish the effect of ultrasound on the electric conductivity of a number of electrolytes used has failed. There are 16 references.

V. Tarisova

[Abstracter's note: Complete translation]

Card 2/2

THE THE PARTY HE STEEL STEEL

TROFIMOV, A.N., inzh.

Current collector of a trolley bus. Vest. elektroprom.
33 no.10:43-47 0 '62.
(Trolley buses) (Electric contactors)

APPROVED FOR RELEASE: 04/03/2001 CIA-RDP86-00513R001756620017-6"

The Market Description and the Assessment for the Company of the C

IVIN, K.V.; MOLODYKH, I.A.; YERMAKOV, N.D.[deceased]; MARKOVNIKOV, V.L., doktor tekhn. nauk; VATSURO, M.A. [deceased]; KRUGLOVA, L.P.; STRAKHOV, K.I.; DUL'KIN, I.A.; FAYN, A.G.; RUBINSKIY, N.V.; SPISKOV, V.S.; PERKIS, D.I., kand. tekhn nauk; LUCHAY, G.A., retsenzent; TROFIMOV, A.N., otv. red. toma; VOLOCHNEV, V.N., red.; SHPOLYANSKIY, M.N., red.; OTOCHEVA, M.A., red.izd-va; LELYUKHIN, A.A., tekhn. red.

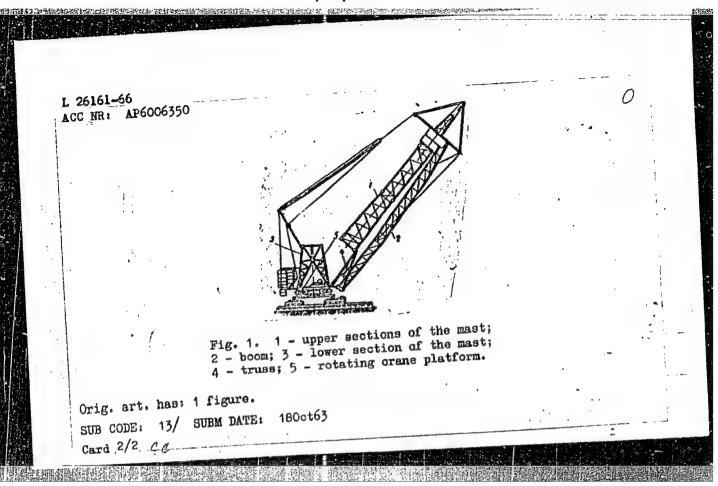
[Technical handbook on electric city transportation in three volumes] Tekhnicheskii spravochnik po gorodskomu elektrotransportu v trekh tomakh. Redkoll.: V.N. Volochnev, A.N. Trofimov, M.N. Shpolianskii. Moskva, Izd-vo M-va Kommun.khoz.RSFSR. Vol. 3. [Trolley buses] Trolleibus. 1963. 722 p. (Trolley buses)

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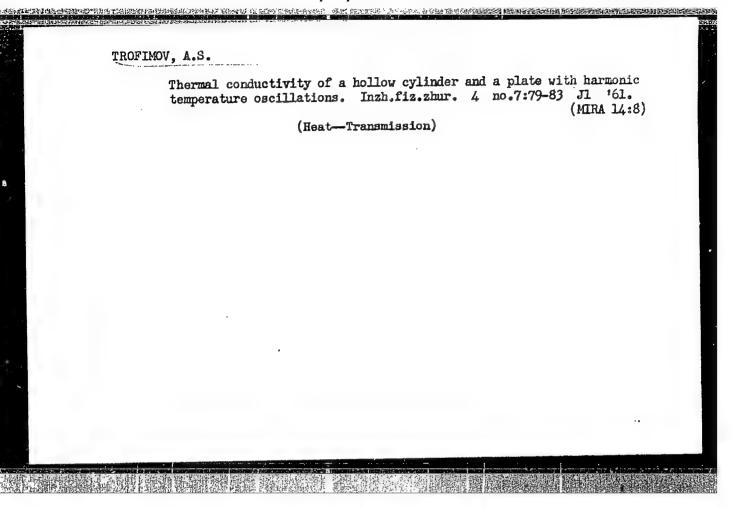
SOURCE CODE: UR/0413/66/000/002/0084/0084 L 26161-66 EWP(h)/EWT(d)/EWP(1)
FCC MR: AP6006350 (A) ACC NR: AP6006350 AUTHORS: Režnik, A. P.; Lobov, A. G.; Auerbakh, V. M.; Trofimov, A. P.; yeshin, K. A.; Vasil'chenko, N. M. ORG: none TITLE: A means of mounting upper sections of crane masts with the boom. Class 35, No. 178071 SOURCE: Izobreteniya, promyshlennyye obraztsy, tovarnyye znaki, no. 2, 1966, 84 TOPIC TAGS: crane, construction equipment ABSTRACT: This Author Certificate describes the mounting of upper sections of crane masts with the boom. The upper sections are set by means of crane mechanisms which are on the lower section of the mast which is on a rotating platform. The leading end of the boom and the base of the supporting part of the mast are joined by a cable which, in turn, is fastened to the edge of the platform. Thus the elevation of the upper sections of the mast is secured by the boom through their turning relative to the place where the truss joins the platform (see Fig. 1). UDC: 621.873.25.002.72 Card 1/2

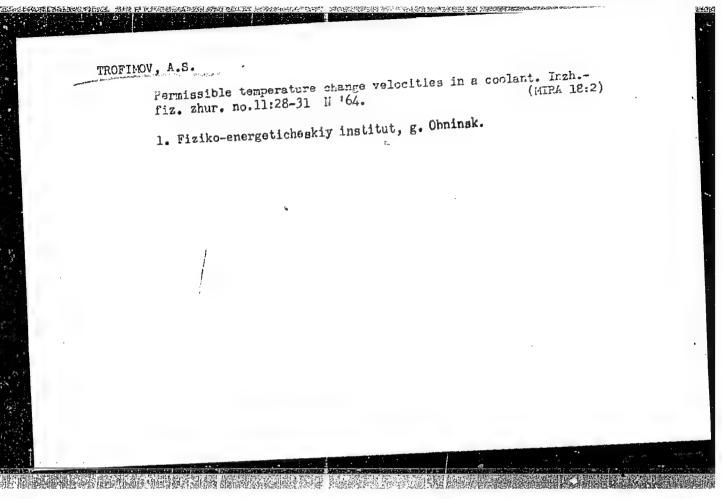


TRAPINOV, A.S.; FOLYAKOV, V.H.

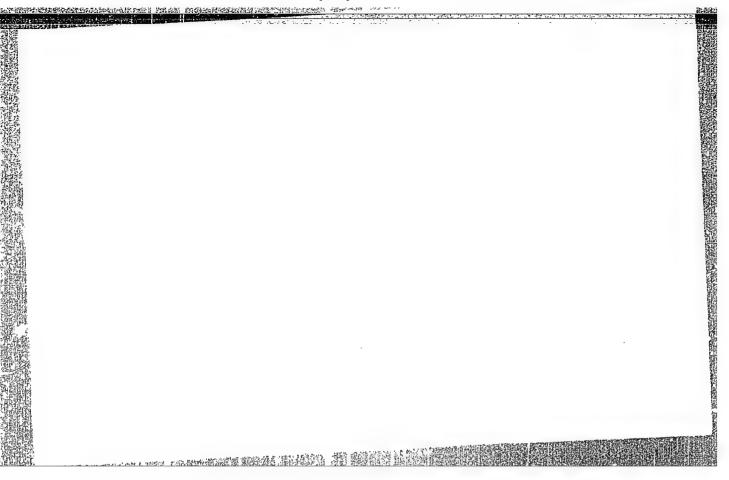
New method of removing used sand in foundries. Lit. proizv. no.1:1116 Ja '58.

(Sand, Foundry)



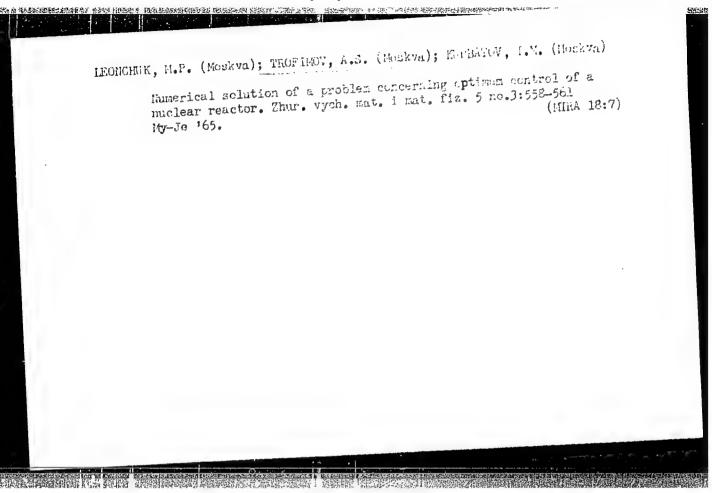






KURPATOV, I.M.; IEONCHUK, M.P.; TROFIMOV, A.S.

Optimum control of thermal processes in a nuclear reactor.
Atom. energ. 19 no.6:537-540 D '65. (MIRA 19:1)



S/170/62/005/004/010/016 B102/B104

262230

AUTHOR:

Trofimov, A. S.

TITLE:

Thermal conductivity of multilayer fuel elements

PERIODICAL: Inzhenerno-fizicheskiy zhurnal, v. 5, no. 4, 1962, 93 - 96

TEXT: A relatively simple approximate method is proposed for calculating the temperature distribution in fuel elements for quasisteady processes. A fuel rod consisting of three layers with  $\rm R_1>R_2>R$  is considered. For

r < R the heat release is assumed to be uniform. The temperatures

$$T(\rho, \tau) = T_{c}(\tau) + (A - \rho^{n}) \varphi(\tau), \quad 0 < \rho < 1;$$

$$T_{2}(\rho, \tau) = T_{c}(\tau) + (B - D\rho^{m}) \varphi(\tau), \quad 1 < \rho < R_{2}/R;$$

$$T_{1}(\rho, \tau) = T_{c}(\tau) + (M - N\rho^{k}) \varphi(\tau), \quad \frac{R_{2}}{R} < \rho < R_{1}/R,$$

$$(1)$$

with

Card (1/3)

S/170/62/005/004/010/016 B102/B104

Thermal conductivity of ...

$$A = \frac{\lambda}{\lambda_1} \frac{n}{k} \left(\frac{R_2}{R}\right)^m \left[\frac{\lambda_1}{\lambda_2} \frac{k}{m} - 1 + \left(\frac{R_1}{R_2}\right)^k \left(1 + k \frac{\lambda_1}{\alpha R_1}\right)\right] - \left(\frac{\lambda}{\lambda_2} \frac{n}{m} - 1\right),$$

$$B = \frac{\lambda}{\lambda_1} \frac{n}{k} \left(\frac{R_2}{R}\right)^m \left[\frac{\lambda_1}{\lambda_2} \frac{k}{m} - 1 + \left(\frac{R_1}{R_2}\right)^k \left(1 + k \frac{\lambda_1}{\alpha R_1}\right)\right],$$

$$D = \frac{\lambda}{\lambda_2} \frac{n}{m},$$

$$M = \frac{\lambda}{\lambda_1} \frac{n}{k} \left(\frac{R_3}{R}\right)^m \left(\frac{R_1}{R_3}\right)^k \left(1 + k \frac{\lambda_1}{\alpha R_1}\right),$$

$$N = \frac{\lambda}{\lambda_1} \frac{n}{k} \left(\frac{R_3}{R}\right)^{m-k}.$$

are to be determined;  $\varrho=r/R$  and  $T_{c}(\tau)$  is the coolant temperature. For  $\varphi(\tau)$ , the equation  $d\varphi/d\tau+\beta\varphi(\tau)=\delta q(\tau)-\xi\,dT_{c}(\tau)/d\tau$  is obtained, where Card 2/3

S/170/62/005/004/010/016 B102/B104

Thermal conductivity of ...

 $q(\tau)$  is the dimensionless heat release. The solution is

$$\varphi(\tau) = \delta \int_{0}^{\tau} q(\tau') e^{-\beta(\tau-\tau')} d\tau' - \xi \int_{0}^{\tau} \frac{dT_{c}(\tau')}{d\tau'} \times$$
(6).

 $\times e^{-\beta(\tau-\tau')} d\tau' + \varphi_0 e^{-\beta\tau}$ 

The formulas obtained are used to calculate a numerical example for a double-layer fuel rod. The results are in satisfactory agreement with data obtained from exact calculations, except for short periods ( $\tau \lesssim 0.05$  sec). There are 1 table and 1 Soviet reference.

SUBMITTED: October 11, 1961

Card 3/3

